



# Preparation and Characterization of Carbon-Filled PVA Composites Produced by Laser Ablation in Liquids

**Dr. Hiroshi Tanaka, PhD**

Advanced Nanomaterials Research Center, Kyoto International University, Kyoto, Japan

## Abstract

*Polymer matrix composites have become indispensable materials across numerous industries due to their tunable mechanical, thermal, and electrical properties. Polyvinyl alcohol (PVA), a water-soluble and biocompatible synthetic polymer, is a versatile matrix material for composite development. The incorporation of carbon-based fillers has long been a strategy to enhance polymer properties, ranging from mechanical reinforcement to electrical conductivity [1]. Recent advances in nanotechnology have introduced novel carbon allotropes, such as carbon nanoparticles, including carbon quantum dots (CQDs) [2, 3], carbon nanodots (CNDs) [4], and graphene quantum dots (GQDs) [5]. These nanoparticles exhibit unique size-dependent optical and electronic properties [2, 3, 4, 5, 6] that differ significantly from traditional carbon fillers like graphite [1]. Laser ablation in liquids (LAL) is a promising method for synthesizing such nanoparticles, producing them directly in a liquid medium suitable for composite fabrication. This article reviews the relevant aspects of PVA as a matrix, the characteristics of advanced carbon nanoparticle fillers based on cited literature [2, 3, 4, 5, 6], considers fabrication methods like LAL for producing such fillers, and discusses the potential properties and applications of resulting PVA-carbon nanoparticle composites, drawing parallels and distinctions with other carbon-filled systems [1].*

## Keywords

*Polyvinyl Alcohol (PVA), Carbon Nanoparticles, Laser Ablation in Liquids (LAL), Polymer Composites, Nanocomposites, Electrical Conductivity, Thermal Stability, Mechanical Properties, Nanomaterials, Composite Fabrication.*

## INTRODUCTION

Polymer composites, materials combining a polymer matrix with reinforcing fillers, offer a broad spectrum of properties tailored for specific applications. The choice of both the polymer and the filler dictates the final characteristics of the composite. Polyvinyl alcohol (PVA) is a synthetic polymer with excellent film-forming capabilities, high tensile strength, chemical resistance, and biocompatibility, making it suitable for diverse applications including drug delivery, tissue engineering, packaging, and sensors. To expand its utility, PVA is often compounded with various fillers.

Carbon materials have historically been used as fillers in polymers to improve properties such as stiffness, strength, electrical conductivity, and thermal conductivity. Traditional carbon fillers include carbon black, carbon fibers, and graphite [1]. The development of nanotechnology has introduced a new class of carbon fillers: carbon nanoparticles. These include carbon quantum dots (CQDs) [2, 3], carbon nanodots (CNDs) [4], graphene quantum dots (GQDs) [5], and other related carbon-based dots [6], typically with dimensions below 10 nanometers. Unlike micron-sized carbon fillers, carbon nanoparticles possess unique quantum confinement effects and surface properties that lead to extraordinary optical and electronic characteristics [2, 3, 4, 5, 6], such as photoluminescence with high quantum yields [5, 6] and excitation-dependent or independent emission [2, 6].

The potential of combining the advantageous properties of PVA with the remarkable characteristics of carbon nanoparticles is significant. Such composites could lead to novel functional materials for areas like flexible electronics, advanced sensors, bioimaging probes, and photocatalysts. Realizing this potential requires effective methods for synthesizing suitable carbon nanoparticles and incorporating them uniformly into the PVA matrix. Laser ablation in liquids (LAL) is a technique that uses pulsed lasers to ablate a solid carbon target submerged in a liquid, directly yielding nanoparticles dispersed in the liquid medium. This method is attractive for creating nanoparticle fillers for PVA, as PVA is water-soluble, potentially allowing for direct mixing

with nanoparticle colloidal suspensions produced by LAL in water.

This article explores the confluence of these elements: PVA as a matrix, the properties of carbon nanoparticles as described in relevant research [2, 3, 4, 5, 6], the suitability of fabrication techniques like LAL for producing such fillers, and the resulting potential properties and applications of PVA-carbon nanoparticle composites. We draw on insights from existing literature on carbon-filled polymers [1] and the specific characteristics of carbon nanoparticles [2, 3, 4, 5, 6] to discuss the opportunities and challenges in this promising area of materials science.

#### Materials and Fabrication Considerations (Adapted Methods Section)

Developing PVA composites filled with carbon nanoparticles involves careful selection of the polymer matrix properties, understanding the characteristics of the specific carbon nanoparticle filler, and choosing an appropriate fabrication method to ensure homogeneous dispersion and maintain nanoparticle integrity and function.

1. **Polymer Matrix: Polyvinyl Alcohol (PVA)** PVA is a semi-crystalline polymer synthesized by the hydrolysis of polyvinyl acetate. Its high density of hydroxyl groups makes it hydrophilic and water-soluble, which is advantageous for processing from aqueous solutions. PVA films exhibit good mechanical strength and flexibility, acting as a robust matrix. Its biocompatibility also makes it suitable for applications in contact with biological systems, including potential bioimaging applications if combined with fluorescent carbon nanoparticles [3]. The choice of PVA grade (molecular weight, degree of hydrolysis) influences its solubility, viscosity in solution, and mechanical properties, all of which affect composite processing and final performance.

2. **Carbon Nanoparticle Fillers** Based on the provided references, several types of carbon nanoparticles are relevant:

- o **Carbon Quantum Dots (CQDs):** These are quasi-spherical carbon nanoparticles typically smaller than 10 nm. References highlight their potential for optical bioimaging [3] and narrow bandwidth emission, making them suitable for applications like colored LEDs [2].

- o **Carbon Nanodots (CNDs):** Similar to CQDs, CNDs are small carbon nanoparticles. They have been explored in hybrid films, such as those combined with ZnO quantum dots, for applications like highly sensitive ultraviolet photodetectors [4].

- o **Graphene Quantum Dots (GQDs):** GQDs are fragments of graphene sheets, typically with lateral dimensions below 100 nm, although the term "quantum dots" usually implies sizes under 10-20 nm. References discuss N-doped GQDs exhibiting high luminescence and excitation-independent emission, investigating their formation mechanism [5].

- o **Other Carbon-Based Dots:** Research also includes co-doping carbon dots with elements like Nitrogen (N) and Sulfur (S) to achieve high quantum yield and excitation-independent emission [6]. These doping strategies are crucial for tailoring the electronic and optical properties of the nanoparticles.

These carbon nanoparticles possess properties distinct from larger carbon fillers like graphite, which is commonly used in polymer composites for applications such as fuel cell components where electrical conductivity is paramount [1]. While graphite flakes enhance conductivity and affect mechanical properties in materials like highly filled polypropylene [1], carbon nanoparticles offer optical properties and potential for quantum effects not typically found in graphite.

3. **Fabrication Methods** Incorporating fillers into a polymer matrix can be achieved through various methods, including melt mixing (suitable for thermoplastics like polypropylene filled with graphite [1]) or solution mixing. For water-soluble PVA, solution mixing is a natural fit, particularly if the carbon nanoparticles are synthesized in an aqueous medium.

**Laser Ablation in Liquids (LAL)** is a technique capable of producing various nanoparticles, including carbon nanoparticles, directly dispersed in a liquid. A high-power pulsed laser beam is focused onto a solid target (e.g., graphite) submerged in a liquid (e.g., water). The laser-induced plasma causes material removal and subsequent nanoparticle formation in the liquid phase. While LAL is a known method for producing carbon nanoparticles, the provided references [2, 3, 4, 5, 6] do not explicitly state that the carbon nanoparticles described within them were synthesized using LAL. However, the output of LAL is a colloidal suspension of nanoparticles in a liquid, which is directly compatible with the solution processing of PVA. This method offers advantages like producing ligand-free nanoparticles and potentially enabling a single-step synthesis and dispersion process for composite fabrication via solution casting. Alternative synthesis methods for carbon nanoparticles mentioned in the literature include chemical oxidation and hydrothermal methods, which also often yield aqueous suspensions suitable for mixing with PVA solutions.

Regardless of the nanoparticle synthesis method, the subsequent composite fabrication step involves mixing the carbon nanoparticle dispersion with the PVA solution and then forming the desired shape (e.g., casting films, spinning fibers, 3D printing). Achieving a uniform dispersion of nanoparticles within the viscous polymer solution and preventing re-agglomeration during solvent removal or curing are critical challenges for maximizing the property enhancement.

#### Potential Properties and Applications (Adapted Results Section)

Based on the known properties of PVA and the characteristics of carbon nanoparticles as highlighted in the provided references, PVA-carbon nanoparticle composites hold promise for a range of functional properties and applications. It is important to note that the cited references focus primarily on the carbon nanoparticles themselves or other composite systems [1], and specific data on PVA composites made with these exact particles by LAL is not provided within this reference set. However, we can infer potential properties based on the filler characteristics.

1. **Optical Properties:** The distinct optical properties of CQDs, CNDs, and GQDs [2, 3, 4, 5, 6] suggest that their incorporation into a transparent PVA matrix could yield luminescent or optically responsive composite materials.

- o CQDs with narrow bandwidth emission [2] could enable PVA films that act as color converters or components in flexible light-emitting devices or displays.
  - o The potential of carbon dots for optical bioimaging [3] suggests that PVA composites could serve as biocompatible fluorescent labels or components in biosensors, assuming the luminescence is retained within the polymer matrix.
  - o N-doped GQDs with high luminescence [5] could enhance the overall brightness and stability of luminescence in the composite.
  - o Carbon dots co-doped with N and S [6] offer excitation-independent emission and high quantum yield, which would be highly desirable features for uniform and efficient luminescence in a composite material, regardless of the excitation source wavelength.
2. **Electronic Properties:** While not typically as conductive as composites filled with carbon black or high aspect ratio carbon fibers or graphite flakes [1], carbon nanoparticle composites can exhibit interesting electronic behaviors.
    - o The use of CNDs in hybrid films for UV photodetectors [4] implies that PVA composites incorporating CNDs could potentially be integrated into optoelectronic devices sensitive to UV light. A PVA-CND layer could serve as a light-absorbing or charge-transporting component within a multi-layer device structure [4].
  3. **Hybrid and Multi-functional Properties:** The concept of hybrid films, such as the ZnO quantum dots/carbon nanodots system for photodetectors [4], illustrates the potential for combining different nanomaterials within a polymer matrix. A PVA composite could potentially serve as a flexible substrate or active layer onto which other nanomaterials are deposited or incorporated to create multi-functional hybrid devices. The versatility of PVA allows for easy processing to form such layered or mixed structures.
  4. **Mechanical and Thermal Properties:** Incorporating fillers into a polymer matrix generally influences mechanical properties like stiffness and strength. While the primary benefit of carbon nanoparticles may be their electronic or optical features, they can also act as reinforcement. The study on highly filled PP/graphite composites [1] highlights the importance of filler content and dispersion on properties like tensile strength and Young's modulus, as well as the impact on processability and surface characteristics relevant for joining [1]. It is reasonable to expect that incorporating carbon nanoparticles into PVA would also affect its mechanical properties, although the scale and nature of the filler (nanoparticle vs. flake) and the matrix (PVA vs. PP) would lead to different outcomes. Thermal properties might also be affected, but specific data would be required.

#### **Challenges and Future Directions (Adapted Discussion Section)**

Despite the exciting potential, fabricating high-performance PVA composites filled with carbon nanoparticles, particularly using methods like LAL, faces several challenges that require dedicated research efforts:

1. **Nanoparticle Dispersion:** Achieving uniform dispersion of carbon nanoparticles in the PVA solution and preventing their re-aggregation during the drying or solidification process is crucial. Nanoparticles have a high surface area and strong van der Waals forces, leading to a tendency to clump together, which can degrade composite properties. Surface modification of nanoparticles or the use of surfactants might be necessary.
2. **Control over LAL Synthesis:** Precisely controlling the size, shape (e.g., triangular CQDs [2]), surface chemistry, and doping level (e.g., N-doped GQDs [5], N,S co-doped dots [6]) of carbon nanoparticles during LAL can be complex. Optimizing laser parameters, target material, and liquid environment is essential to reliably produce nanoparticles with desired properties for specific applications [2, 5, 6].
3. **Property Characterization:** Thorough characterization of the resulting composite material is needed. This includes evaluating the success of nanoparticle dispersion (e.g., microscopy), quantifying the optical properties (luminescence intensity, wavelength, quantum yield) within the polymer matrix, measuring electronic conductivity or semiconducting behavior, and assessing mechanical and thermal properties. The methodology for characterizing filled polymer composites, as discussed in studies like that on PP/graphite [1], provides a framework, but adaptations are needed for nanoscale fillers and optical properties.
4. **Interfacial Interaction:** The interaction between the carbon nanoparticles and the PVA matrix significantly influences load transfer (for mechanical properties) and energy transfer (for optical/electronic properties). Understanding and optimizing the nanoparticle-polymer interface is key.
5. **Scalability and Cost:** Scaling up the production of carbon nanoparticles via LAL and the subsequent composite fabrication process to meet industrial demands remains a challenge. Cost-effectiveness compared to traditional materials must be demonstrated.
6. **Stability and Durability:** Ensuring the long-term stability of the composite's properties, particularly luminescence [5, 6], under various environmental conditions (humidity, UV exposure, temperature) is vital for practical applications.

#### **Future research directions in this field are promising:**

- Developing advanced LAL techniques or other synthesis methods tailored to produce specific carbon nanoparticle types (CQDs, CNDs, GQDs) with precisely controlled properties for incorporation into PVA [2, 3, 4, 5, 6].
- Exploring novel composite fabrication techniques, perhaps integrating LAL directly with a continuous PVA processing method.
- Conducting systematic studies on the effect of nanoparticle concentration, size, doping [5, 6], and surface modification on the mechanical, thermal, optical, and electronic properties of PVA composites.
- Investigating specific applications suggested by the filler properties, such as flexible displays (based on [2]), advanced

bioimaging probes (based on [3]), flexible photodetectors (based on [4]), or components in wearable electronics.

- Exploring the creation of hierarchical or multi-phase composites, perhaps combining different types of fillers or hybridizing the PVA-carbon nanoparticle matrix with other functional materials [4].

## CONCLUSION

Polyvinyl alcohol offers a versatile, water-soluble matrix well-suited for composite fabrication. The emergence of carbon nanoparticles, including CQDs, CNDs, and GQDs [2, 3, 4, 5, 6], provides a new class of fillers with remarkable optical and electronic properties distinct from conventional carbon materials like graphite [1]. While this specific composite fabricated via LAL is an area requiring further dedicated exploration, the component materials and nanoparticle synthesis methods suggest significant potential. Techniques like LAL are promising for producing these nanoparticles directly in liquids, compatible with PVA processing. Leveraging the unique luminescence [2, 5, 6] and electronic behavior [4] of these nanoparticles within a PVA matrix could lead to novel materials for bioimaging [3], flexible electronics, and sensing applications. Overcoming challenges related to nanoparticle dispersion, synthesis control, and comprehensive composite characterization will be crucial to unlock the full potential of PVA composites reinforced with advanced carbon nanoparticles, paving the way for next-generation functional materials.

## REFERENCES

1. Rzeczkowski, Piotr, Beate Krause and Petra Pötschke. "Characterization of highly filled pp/graphite composites for adhesive joining in fuel cell applications." *Polymers* 11 (2019): 462. Google Scholar, Crossref, Indexed at
2. Yuan, Fanglong, Ting Yuan, Laizhi Sui and Zhibin Wang, et al. "Engineering triangular carbon quantum dots with unprecedented narrow bandwidth emission for multicolored leds." *Nat Commun* 9 (2018): 2249. Google Scholar, Crossref, Indexed at
3. Luo, Pengju G., Sushant Sahu, Sheng-Tao Yang and Sumit K. Sonkar, et al. "Carbon "quantum" dots for optical bioimaging." *J Mater Chem B* 1 (2013): 2116-2127. Google Scholar, Crossref, Indexed at
4. Guo, Deng-Yang, Chong-Xin Shan, Song-Nan Qu and De-Zhen Shen. "Highly sensitive ultraviolet photodetectors fabricated from ZnO quantum dots/carbon nanodots hybrid films." *Sci Rep* 4 (2014): 7469. Google Scholar, Crossref, Indexed at
5. Qu, Dan, Min Zheng, Ligong Zhang and Haifeng Zhao, et al. "Formation mechanism and optimization of highly luminescent n-doped graphene quantum dots." *Sci Rep* 4 (2014): 5294. Google Scholar, Crossref, Indexed at
6. Dong, Yongqiang, Hongchang Pang, Hong Bin Yang and Chunxian Guo, et al. "Carbon-based dots co-doped with nitrogen and sulfur for high quantum yield and excitation-independent emission." *Angew Chem Int Ed* 52 (2013): 7800-7804.