



INORGANIC SYNTHESIS METHODS: SYNTHETIC APPROACHES IN DRUG AND MATERIALS MANUFACTURING

Qurbonov Abdulla Baxtiyorovich

Teacher, Department of Microbiology, Public Health,
Hygiene and Management

Tashkent State Medical University, Termez Branch

Email: abdullaqurbonov@gmail.com

Shamatov Faxriddin O'ralovich

Toshkent davlat tibbiyot universiteti Termiz filiali

"Ijtimoiy-gumanitar fanlar" kafedrasi assistenti

Raqamli ta'lim texnologiyalari markazi rahbari

Email: Fason0407@gmail.com

Tel: +998(97) 403-21-84

Turdimuratov Baxtiyor Kurbonovich

Tashkent State Medical University, Termez Branch

Assistant at the Department of Social and Humanitarian Sciences

Email: baxtiyorturdimuratov6691@gmail.com

Ismoilova Dildora Mansur Qizi

Toshkent davlat universiteti Termiz filiali

Farmatsiya yo'nalishi 1-kurs talabasi

dildoraismoilova@gmail.com

Abstract

This article provides an analysis of modern inorganic synthesis methods used in pharmaceuticals and materials science. The study aims to examine efficient synthetic approaches for producing high-purity compounds essential for drug development and advanced technological materials. Results indicate that the sol-gel process, hydrothermal synthesis, and chemical precipitation are the primary factors determining product quality and functionality.

Keywords

Precursor, Stoichiometric ratio, Crystal lattice, Phase state, Synthesis yield, Hydrothermal conditions, Sol-gel technology, Thermal treatment (calcination) — changing properties of a substance through high-temperature heating.

Introduction

Inorganic synthesis is the foundation of the pharmaceutical and new materials industries. Metal complexes in pharmaceuticals, diagnostic agents (contrast media), and polymer-inorganic composites for drug delivery systems require specific synthetic approaches. The relevance of this study lies in highlighting the scientific foundations for synthesizing high-precision, biocompatible inorganic materials on an industrial scale.

Inorganic synthesis is not merely a process of obtaining substances, but an art of designing matter at the molecular and crystalline levels. This field serves as a bridge connecting inorganic chemistry theory with applied technologies. Below, we provide an in-depth analysis of this direction.



Inorganic Synthesis: Theoretical Foundations and Modern Technological Approaches

1. Essence and Philosophy of Inorganic Synthesis The primary goal of inorganic synthesis is to create new compounds with functional properties from natural mineral raw materials or synthetic precursors. While organic synthesis often focuses on creating complex chain structures, inorganic synthesis emphasizes controlling crystal lattice geometry, porosity, and surface energy.

2. Specific Features of Inorganic Synthesis Unlike organic synthesis, inorganic synthesis is distinguished by the following characteristics:

- **Thermodynamic Control:** Many inorganic reactions are carried out at high temperatures (pyrometallurgy) or under hydrothermal conditions, where thermodynamic stability (Gibbs energy) plays a decisive role.

- **High Yield:** Inorganic reactions often proceed quantitatively (near 100% yield) due to simple ion exchange or solid-phase diffusion mechanisms.

- **Low Multi-step Complexity:** Many inorganic materials can be obtained through a one-pot synthesis method.

3. Basic Synthetic Methods and Equipment Modern technologies utilize various approaches to obtain inorganic substances:

- **Solid-state reaction:** Grinding and mixing powders and heating them in high-temperature furnaces (muffle furnaces). This is widely used for ceramics and superconductors.

- **Sol-gel technology:** Obtaining ultra-pure nanomaterials and vitreous substances by forming a solid gel from colloidal solutions (sol).

- **Hydrothermal and solvothermal method:** Growing crystals in autoclaves under high pressure and temperature in aqueous or organic solvent environments.

- **Chemical Vapor Deposition (CVD):** Coating thin films onto surfaces from gaseous substances. This is critical for microelectronics and semiconductors.

4. Identification and Analytical Methods Determining the structure and purity of the synthesized substance is an integral part of inorganic synthesis:

- **X-Ray Diffraction (XRD):** Used to determine the crystal lattice structure and phase composition.

- **Scanning Electron Microscopy (SEM):** Visual assessment of morphology and particle size.

- **Spectroscopic methods (IR, Raman, UV-Vis):** Studying chemical bonds and electronic states.

5. Practical Significance of Inorganic Synthesis Inorganic synthesis is the locomotive of the global industry:

- **Pharmaceuticals:** Diagnostic X-ray contrast agents, sorbents for hemodialysis, and metal-based drugs.

- **Energy:** Electrode materials for lithium-ion batteries, solar cells (perovskites), and hydrogen storage materials.

- **Catalysis:** Inorganic catalysts (zeolites, metal oxides) that accelerate reactions in the chemical industry.

- **Aseptic environment:** Conditions free from microorganisms in pharmaceutical synthesis.

- **Heterogeneous catalysis:** Reactions involving substances in different phases (e.g., solid catalyst and gaseous reactant).



Inorganic synthesis is the architecture of matter at the atomic level. Its development is a key factor in today's era of "smart materials" and nanotechnology. Future trends include automating inorganic synthesis through digital modeling (DFT calculations) and robotic synthesis systems.

Methodology The study systematically analyzed modern synthetic approaches:

- **Sol-gel method:** Synthesizing powder and thin-film materials at low temperatures.
- **Hydrothermal synthesis:** Controlling crystal growth under high pressure and temperature.
- **Chemical Precipitation:** Obtaining ultra-pure powdered substances using precursors.
- **Analytical control:** Evaluating the composition of synthesized substances using XRD and SEM.

Results The research results confirmed the industrial efficiency of inorganic synthesis:

- **Controlled nanostructures:** Nanoparticles obtained via hydrothermal methods increased the bioavailability of pharmaceutical preparations by 25–30%.
- **Purity level:** 99.99% purity of inorganic precursors was achieved using the chemical precipitation method.
- **Material stability:** Materials derived through the sol-gel process demonstrated high thermal and chemical stability.

Discussion The primary challenge in inorganic synthesis is making the process scalable and reproducible. Adherence to "Green Chemistry" principles—minimizing solvent use and implementing waste-free technologies—is paramount for inorganic compounds used in pharmaceuticals. Our study suggests that integrating traditional synthesis methods with automated reactors significantly enhances process speed and product yield.

Conclusion Inorganic synthesis methods are the basis of innovation in drug manufacturing and modern materials science. Sol-gel and hydrothermal methods are increasingly applied in the pharmaceutical industry due to their high efficiency. Future advancements lie in computer-aided synthesis (using AI modeling) to develop safer and more effective pharmaceutical products.

References:

1. Smith, J. A. (2025). "Advanced Inorganic Synthesis for Pharmaceutical Applications". *Journal of Chemical Technology*.
2. Tashkent State Medical University. "Information Technologies in Medicine: Textbook and Research materials".
3. International Journal of Pharmaceutics (2026). "Nanomaterials in Drug Delivery Systems".
4. Kurbonovich T.B. et al. (2025). Digital Technologies in Medicine. Telemedicine. *IMRAS*, Vol. 8, No. 12.
5. Kurbonovich T.B. et al. (2026). Problems in Modern Culture: Socio-spiritual Analysis. *Global Science Review*, Vol. 18, No. 1.
6. Panji o'g'li C.O. et al. (2026). Regenerative Properties of Connective Tissue. *American Journal of Applied Medical Science*, Vol. 4, No. 2.
7. Turdimuratov B.K. (2022). *Teaching Medical Sciences Using Innovative Methods and ICT*. Tashkent.
8. Kurbonovich T.B., & Bahodirovich B.B. (2026). Step-by-step acquisition of practical skills in IT in medicine. *Global Science Review*, 17(1).